

# Isolation of the Fumonisin Mycotoxins: A Quantitative Approach

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A method for the preparative-scale isolation of the fumonisin B (FB) mycotoxins, from corn cultures of *Fusarium moniliforme*, is described and quantitatively evaluated. Eighty percent of FB<sub>1</sub> and 60% of FB<sub>2</sub> were recovered after extraction with CH<sub>3</sub>OH/H<sub>2</sub>O (3:1). The fumonisins, including the newly discovered FB<sub>3</sub> and FB<sub>4</sub>, were purified using Amberlite XAD-2, silica gel, and reverse-phase C<sub>18</sub> chromatography. The Amberlite XAD-2 purification step proved to be the most effective cleanup procedure, while subsequent chromatography on silica gel and RP C<sub>18</sub> effectively separate the individual fumonisins to a purity of over 90%. The relatively low final yield (40%) of FB<sub>1</sub> and FB<sub>2</sub> may be ascribed to (1) the strong affinity of FB<sub>1</sub> for silica gel, (2) the low initial recovery (60%) of FB<sub>2</sub>, and (3) the formation of monomethyl and dimethyl esters of FB<sub>1</sub> and FB<sub>2</sub>, as well as their interference in the purification of the individual fumonisins. The *N*-acetyl derivatives of FB<sub>1</sub> and FB<sub>2</sub> were also purified and shown to be metabolites of *F. moniliforme*.

## INTRODUCTION

Recent investigations on the mycotoxins produced by *Fusarium moniliforme* Sheldon indicate that the fumonisin mycotoxins could be responsible for the major toxicological and carcinogenic effects caused by corn cultures of the fungus in experimental animals (Gelderblom et al., 1988a,b, 1991; Kellerman et al., 1990; Marasas et al., 1988; Voss et al., 1989). At present little is known about the biological effects of these toxins in different animal species. The need to develop efficient and cost-effective methods for purifying sufficient quantities for biological evaluation is therefore of particular importance since the fumonisin mycotoxins occur under natural conditions (Sydenham et al., 1990a,b; Shephard et al., 1990) and could enter the human and animal food chains.

The first paper on the isolation and purification of fumonisin B<sub>1</sub> (FB<sub>1</sub>) and fumonisin B<sub>2</sub> (FB<sub>2</sub>) dealt mainly with the detection of cancer-promoting compounds from cultures of a carcinogenic strain of *F. moniliforme*; the purification techniques were described only qualitatively (Gelderblom et al., 1988a). Extraction of the fumonisin mycotoxins was achieved with CH<sub>3</sub>OH/H<sub>2</sub>O (3:1) followed by a solvent-partitioning step using CHCl<sub>3</sub>. The subsequent purification of the aqueous phase was effected on Amberlite XAD-2, silica gel, and reverse-phase (C<sub>18</sub>) chromatographic columns yielding FB<sub>1</sub> and FB<sub>2</sub>, both with a purity of approximately 90%.

This paper describes methods for the quantitative purification of FB<sub>1</sub>, FB<sub>2</sub>, and the newly discovered fumonisin B<sub>3</sub> (FB<sub>3</sub>) from corn cultures of *F. moniliforme* strain MRC 826. In addition, several other structurally related compounds, including fumonisin A<sub>1</sub> (FA<sub>1</sub>) and fumonisin A<sub>2</sub> (FA<sub>2</sub>), which were previously observed during the purification of FB<sub>1</sub> and FB<sub>2</sub> (Gelderblom et al., 1988a; Bezuidenhout et al., 1988), and fumonisin B<sub>4</sub> (FB<sub>4</sub>) were purified and their chemical structures determined.

## MATERIALS AND METHODS

**Chemicals.** Amberlite XAD-2 and silica gel 60 (Kieselgel 60, 0.063–0.200 mm) were purchased from Merck SA. The reverse-phase (RP) packing material (C<sub>18</sub>; Bondesil; 40 μm) was obtained from Analytichem International, Harbour City, CA, or prepared from silica gel 60 (0.063–0.200 mm) according to the method of Kingston and Gerhart (1976). All of the organic solvents were of analytical grade and glass-distilled prior to use during the final column purification steps on silica gel and RP C<sub>18</sub>.

**Fungal Cultures.** Corn cultures of *F. moniliforme* strain 826 were prepared as described previously (Alberts et al., 1990) and incubated in the dark for 6 weeks at 25 °C. Cultures were oven-dried at 50 °C, ground, and stored at 4 °C until required.

**Fumonisin Standards.** FB<sub>1</sub>, FB<sub>2</sub>, and FB<sub>3</sub> (Figure 1) were purified according to the method described in the present paper. For use as analytical standards the fumonisins were further subjected to two successive column separations on silica gel (second column) and reverse-phase (C<sub>18</sub>) columns as described below. The purity of both standards was verified by nuclear magnetic resonance (NMR), high-performance liquid chromatography (HPLC) (Alberts et al., 1990), and silica gel thin-layer chromatography (TLC), and they are considered for the purpose of this study as 100% pure.

**Detection and Quantification of the Fumonisin.** All of the sample extracts and column fractions were analyzed by silica gel TLC for the presence of the different fumonisins using CHCl<sub>3</sub>/CH<sub>3</sub>OH/H<sub>2</sub>O/CH<sub>3</sub>COOH (55:36:8:1) and/or CHCl<sub>3</sub>/CH<sub>3</sub>OH/CH<sub>3</sub>COOH (6:3:1) as developing solvents. The TLC plates were developed by spraying with a *p*-anisaldehyde [0.5 g of *p*-anisaldehyde in CH<sub>3</sub>OH/CH<sub>3</sub>COOH/H<sub>2</sub>SO<sub>4</sub> (8:10:5)] and/or a ninhydrin (0.2% in ethanol) solution and heated at 120 °C until color development. On the basis of the TLC analyses, column fractions were combined in such a manner that the most effective separation between the individual fumonisins was accomplished. Subsamples of the combined column fractions and sample extracts were kept throughout the purification process for the quantification of FB<sub>1</sub>, FB<sub>2</sub>, and FB<sub>3</sub>.

Quantification of the fumonisins in the sample extracts and combined column fractions was achieved by HPLC analyses of the methyl derivatives as described previously (Alberts et al., 1990). The mobile phases used consisted of (1) CH<sub>3</sub>OH/0.1 M KH<sub>2</sub>PO<sub>4</sub> (7:3; 10 mM MgCl<sub>2</sub>; pH 3.5) for FB<sub>1</sub> analyses and (2) CH<sub>3</sub>OH/0.1 M KH<sub>2</sub>PO<sub>4</sub> (18:7; 10 mM MgCl<sub>2</sub>; pH 3.5) for FB<sub>2</sub> and FB<sub>3</sub> analyses.

**Extraction and Solvent Partitioning.** Culture material (1 kg) was extracted twice with ethyl acetate (1 L) by blending (Waring blender) and filtering (Whatman No. 4). The residue

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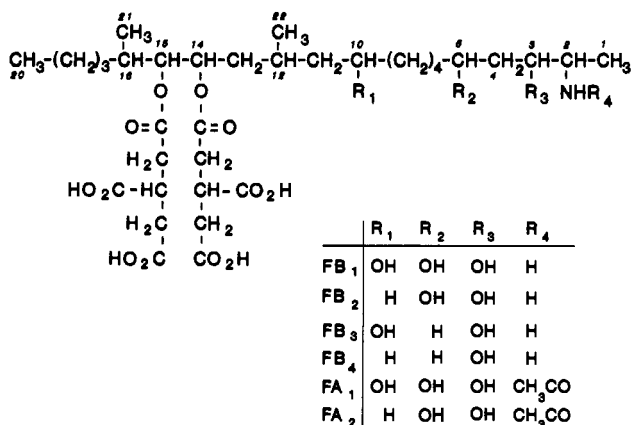
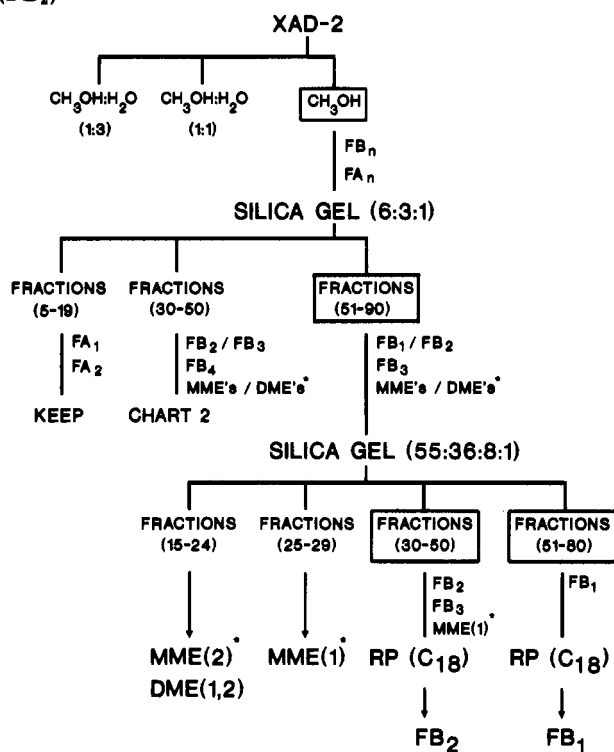


Figure 1. Chemical structures of the different fumonisins produced by *F. moniliforme*.

Chart I. Schematic Representation of the Procedures Used for the Isolation of Fumonisin B<sub>1</sub> (FB<sub>1</sub>) and B<sub>2</sub> (FB<sub>2</sub>)



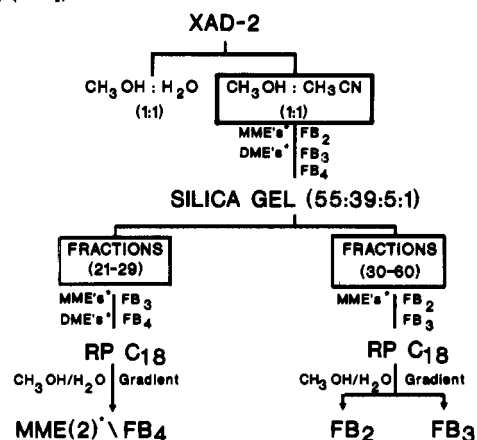
\* MMEs/DMEs, monomethyl esters/dimethyl esters; MME(1), monomethyl ester of FB<sub>1</sub>; MME(2), monomethyl ester of FB<sub>2</sub>; DME(1,2), dimethyl ester of FB<sub>1</sub> and FB<sub>2</sub>, respectively.

was extracted once with 1 L and twice with 1.5 L of CH<sub>3</sub>OH/H<sub>2</sub>O (3:1) as described above. The combined extracts were evaporated to dryness under vacuum at 50 °C; the residual material was dried at 60 °C and stored at 4 °C for FB<sub>1</sub> and FB<sub>2</sub> analyses. The dried culture extract was dissolved in CH<sub>3</sub>OH/H<sub>2</sub>O (1:3, 200 mL) at 50 °C and partitioned with CHCl<sub>3</sub> (3 × 100 mL) as described previously (Gelderblom et al., 1988a). The aqueous phase was evaporated to dryness under vacuum at 50 °C. A schematic diagram of the column chromatographic purification procedures for the different fumonisins is outlined in Charts I and II.

**Purification of FB<sub>1</sub>.** *Amberlite XAD-2.* The above extraction residue (125 ± 32 g, Table I) was dissolved in CH<sub>3</sub>OH/H<sub>2</sub>O (1:3; 100 mL) and applied to an Amberlite XAD-2 column (7 × 87 cm; sample:resin 1:15) previously equilibrated with CH<sub>3</sub>OH/H<sub>2</sub>O (1:3). After the column was washed successively with CH<sub>3</sub>OH/H<sub>2</sub>O (1:3; 1 L) and CH<sub>3</sub>OH/H<sub>2</sub>O (1:1; 1.5 L), the fumonisin mycotoxins were eluted with CH<sub>3</sub>OH (1 L).

*Silica Gel 60. First Column [CHCl<sub>3</sub>/CH<sub>3</sub>OH/CH<sub>3</sub>COOH (6:3:1) as Eluant].* The residue obtained from the methanol fraction

Chart II. Schematic Representation of the Procedures Used for the Isolation of Fumonisin B<sub>2</sub> (FB<sub>2</sub>), B<sub>3</sub> (FB<sub>3</sub>), and B<sub>4</sub> (FB<sub>4</sub>)



\* MMEs, monomethyl esters, DMEs, dimethyl esters; MME(2), monomethyl ester of FB<sub>2</sub>.

Table I. Quantification of the Extraction of FB<sub>1</sub> and FB<sub>2</sub> from Fungal Culture Material

sample	wt, g	FB <sub>1</sub> concn, <sup>a</sup> g	FB <sub>2</sub> concn, <sup>a</sup> g
corn culture	1000	2.24 ± 0.40	1.12 ± 0.33
ethyl acetate	81.1 ± 3.4	nd <sup>b</sup>	nd <sup>b</sup>
CH <sub>3</sub> OH/H <sub>2</sub> O (3:1)	125.2 ± 17.6	1.82 ± 0.5 (81.3%) <sup>c</sup>	0.67 ± 0.2 (59.8%) <sup>c</sup>

<sup>a</sup> Values represent means ± standard deviation of two different samples each done in triplicate. <sup>b</sup> nd, not detected. <sup>c</sup> The percentage of FB<sub>1</sub> and FB<sub>2</sub> recovered in the CH<sub>3</sub>OH/H<sub>2</sub>O (3:1) fraction is given in parentheses.

(10 ± 0.6 g) was first fractionated on a silica gel column (5.5 × 85 cm; 1100 g) using CHCl<sub>3</sub>/CH<sub>3</sub>OH/CH<sub>3</sub>COOH (6:3:1) as the mobile phase. The sample was dissolved in the eluant (70 mL) and applied to the column which contained anhydrous Na<sub>2</sub>SO<sub>4</sub> (ca. 50 g) on top of the silica gel. After elution of 1.3 L of mobile phase at a flow rate of 3.5 mL/min, fractions (50 mL) were collected and analyzed by TLC. Fractions containing FA<sub>1</sub> and FA<sub>2</sub> (5-19), FB<sub>2</sub> and FB<sub>3</sub> (30-50), and FB<sub>1</sub>, FB<sub>2</sub>, and FB<sub>3</sub> (51-90) were combined separately (Chart I) and the solvents evaporated at 50 °C under reduced pressure. The subsequent purification of the FB<sub>2</sub>/FB<sub>3</sub> mixture (fractions 30-50) and the other structurally related compounds is described separately.

*Second Column [CHCl<sub>3</sub>/CH<sub>3</sub>OH/H<sub>2</sub>O/CH<sub>3</sub>COOH (55:36:8:1) as Eluant].* The main FB<sub>1</sub> fraction (51-90; 4.8 ± 0.22 g) was purified by silica gel column chromatography as described above, but without the Na<sub>2</sub>SO<sub>4</sub>, using CHCl<sub>3</sub>/CH<sub>3</sub>OH/H<sub>2</sub>O/CH<sub>3</sub>COOH (55:36:8:1) as eluant. The sample, dissolved in 20 mL of the eluant, was applied to the column and eluted at 2 mL/min. After elution of 1.2 L of mobile phase, the flow rate was changed to 3.5 mL/min while fractions (50 mL) were collected and analyzed by TLC. Fractions 50-80 (2.14 ± 0.32 g) contained only FB<sub>1</sub>. The fractions containing a mixture of FB<sub>2</sub>, FB<sub>3</sub>, and other related compounds, such as monomethyl esters (MMEs) and dimethyl esters (DMEs), were combined separately (Chart I).

*Reverse Phase (C<sub>18</sub>).* The final purification of FB<sub>1</sub> was achieved on a RP C<sub>18</sub> column (1.5 × 50 cm; 200 g) equipped with end fittings and using CH<sub>3</sub>OH/H<sub>2</sub>O (1:1) as eluant. The eluant flow was maintained at 1.5 mL/min with the aid of a peristaltic pump. The pH of the sample, dissolved in CH<sub>3</sub>OH/H<sub>2</sub>O (1:1; 20 mL), was adjusted to 3.5 with 1 N HCl prior to application to the column. Fractionation was effected by running a 800-mL linear gradient from CH<sub>3</sub>OH/H<sub>2</sub>O (1:1) to CH<sub>3</sub>OH/H<sub>2</sub>O (4:1) at a flow rate of 1.5 mL/min. Fractions (15 mL) were collected after 400 mL of the gradient had eluted from the column. The purity of FB<sub>1</sub> (1.33 ± 0.44 g), which eluted as a single peak from the column (fractions 8-15), was determined by HPLC (Alberts et al., 1990).

*Purification of FB<sub>2</sub>. Silica Gel.* During the first chromatographic separation on silica gel the FB<sub>3</sub>-containing fractions (30-50) and the FB<sub>1</sub>-containing fractions (51-90) each contained

similar amounts of FB<sub>2</sub>. The purification of FB<sub>2</sub> from the FB<sub>2</sub>/FB<sub>3</sub> sample will be described separately. The FB<sub>2</sub> that coeluted with FB<sub>1</sub> and some FB<sub>3</sub> was fractionated on a silica gel column as described above (second column). As indicated in Chart I, FB<sub>2</sub> eluted first from the column (fractions 30–50) followed by FB<sub>1</sub>, while a small amount of FB<sub>3</sub> coeluted with FB<sub>2</sub>.

**Reverse Phase (C<sub>18</sub>).** The C<sub>18</sub> material, prepared from silica gel as described by Kingston and Gerhart (1976), was equilibrated with CH<sub>3</sub>OH/H<sub>2</sub>O (1:1) and packed in a column (3 × 27 cm) supplied with end fittings. The sample, at a pH of 3.5, was fractionated using a 800-mL linear gradient from CH<sub>3</sub>OH/H<sub>2</sub>O (1:1) to CH<sub>3</sub>OH while the flow was maintained at 1.5 mL/min. Fractions (15 mL) were collected after 300 mL had eluted from the column. FB<sub>2</sub>, with a purity of 90%, was obtained in fractions 17–30. FB<sub>3</sub>, which coeluted with FB<sub>2</sub> during the course of the second silica gel column fractionation, was obtained in fractions 5–14 with a purity of only 20%.

**Purification of FB<sub>3</sub>.** Amberlite XAD-2. The material containing FB<sub>2</sub> and FB<sub>3</sub> (fractions 30–50, Chart I), obtained from the first silica gel column purification, was further fractionated on Amberlite XAD-2 (Chart II). The Amberlite XAD-2 column (4.5 × 30 cm) was equilibrated with CH<sub>3</sub>OH/H<sub>2</sub>O (1:1), and the sample (3.24 ± 0.2 g), dissolved in the same solvent, was applied after the pH had been adjusted to 3.5 with 0.1 N HCl. After the column was washed with 400 mL of CH<sub>3</sub>OH/H<sub>2</sub>O (1:1), the fumonisins were eluted with CH<sub>3</sub>OH/CH<sub>3</sub>CN (1:1; 250 mL). Fractions containing both FB<sub>2</sub> and FB<sub>3</sub> were combined and the solvents evaporated in vacuo at 50 °C.

**Silica Gel 60.** A silica gel column (5 × 45 cm; 300 g) was equilibrated with CHCl<sub>3</sub>/CH<sub>3</sub>OH/H<sub>2</sub>O/CH<sub>3</sub>COOH (55:39:5:1). The material obtained from the Amberlite XAD-2 column (2.16 ± 0.20 g) was applied to the column, and after elution of 450 mL (1.0 mL/min), fractions (30 mL) were collected. Fractions 30–60, containing a mixture of FB<sub>2</sub> and FB<sub>3</sub>, were pooled and the solvents evaporated at 50 °C.

**Reverse Phase (C<sub>18</sub>).** Final separation of FB<sub>2</sub> and FB<sub>3</sub> was achieved on the RP C<sub>18</sub> column described above. The column was first regenerated by successive elution with CH<sub>3</sub>OH (250 mL) containing 0.1 M MgCl<sub>2</sub> and H<sub>2</sub>O (250 mL) and subsequently equilibrated with CH<sub>3</sub>OH/H<sub>2</sub>O (1:1). After application of the sample (1.99 ± 0.10 g, pH 3.5), separation was achieved by gradient elution as described earlier using a 800-mL gradient from CH<sub>3</sub>OH/H<sub>2</sub>O (1:1) to CH<sub>3</sub>OH. FB<sub>3</sub> (0.38 ± 0.02 g), which eluted first, chromatographed as a single spot on TLC and was subjected to further analyses (HPLC and NMR) to confirm its purity. FB<sub>2</sub> (0.37 ± 0.01 g) with a purity of 92% was obtained from subsequent fractions.

**Purification of Structurally Related Compounds.** In the course of purifying FB<sub>1</sub>, FB<sub>2</sub>, and FB<sub>3</sub> several other compounds that reacted similarly to the fumonisins with both *p*-anisaldehyde and ninhydrin were observed by silica gel TLC. On the basis of their reaction with ninhydrin these compounds were divided into ninhydrin-positive and ninhydrin-negative compounds.

**Ninhydrin Positive Compounds.** TLC analysis of the CH<sub>3</sub>OH fraction from the Amberlite XAD-2 column and the fractions obtained during its subsequent purification on silica gel (first column) indicated the presence of several spots from compounds with higher *R<sub>f</sub>* values than FB<sub>1</sub> and FB<sub>2</sub> (Table IV). These spots were eventually shown to consist of a mixture of the monomethyl and dimethyl esters of both FB<sub>1</sub> and FB<sub>2</sub> and another fumonisin given the trivial name of FB<sub>4</sub> (Figure 1). The same fractionation steps used during the purification of FB<sub>2</sub> and FB<sub>3</sub> were applied for the isolation of FB<sub>4</sub> (Chart I). Subsequent fractionation of FB<sub>4</sub>, after the RP C<sub>18</sub> purification step, was performed on a silica gel column using CHCl<sub>3</sub>/CH<sub>3</sub>OH/H<sub>2</sub>O/CH<sub>3</sub>COOH (55:36:8:1) as eluant. The FB<sub>4</sub> was finally fractionated through a Sep-Pak C<sub>18</sub> cartridge as described below and subjected to structural analyses.

After the second silica gel column, a pure preparation of a mixture of monomethyl esters (MMEs) of FB<sub>1</sub> containing some inorganic compounds was obtained (Chart I). The fraction containing the mixture of DMEs of FB<sub>1</sub> and FB<sub>2</sub> and the MME of FB<sub>2</sub> was further fractionated on a silica gel column (2 × 60 cm) using CHCl<sub>3</sub>/CH<sub>3</sub>OH/H<sub>2</sub>O/CH<sub>3</sub>COOH (55:36:8:1) as the eluant. This chromatographic system achieved good separation

between the DME of FB<sub>1</sub> and the MME and DME of FB<sub>2</sub>. Final purification of the MMEs and DMEs of FB<sub>1</sub> and FB<sub>2</sub> for structure elucidation by NMR was achieved on a C<sub>18</sub> Sep-Pak cartridge (Waters, Millipore Corp., Milford, MA). The cartridge was washed with CH<sub>3</sub>OH and equilibrated with CH<sub>3</sub>OH/H<sub>2</sub>O (1:1). After application of the sample (pH 3.5), 20 mL of the eluant was passed through the cartridge. The compounds were eluted using CH<sub>3</sub>OH. The solvent was evaporated and the pure compounds subjected to structural analyses.

**Ninhydrin Negative Compounds.** Fractions obtained during chromatography on the first silica gel column revealed two spots on TLC with *R<sub>f</sub>* values of 0.38 and 0.46, respectively, upon spraying with *p*-anisaldehyde (Table IV). These compounds, called FA<sub>1</sub> and FA<sub>2</sub> (Bezuidenhout et al., 1988) (Figure 1), were further purified on an Amberlite XAD-2 column (4.5 × 30 cm) using the same elution procedures as described for the purification of FB<sub>3</sub>. The material, still containing both FA<sub>1</sub> and FA<sub>2</sub>, was subsequently fractionated on a silica gel column (2 × 65 cm) using CHCl<sub>3</sub>/CH<sub>3</sub>OH/CH<sub>3</sub>COOH/H<sub>2</sub>O (65:25:6:4) as eluant. TLC analyses of the fractions indicated that FA<sub>1</sub> (fractions 30–40) and FA<sub>2</sub> (fractions 20–27) were cleanly separated and each combined fraction exhibited a single spot on TLC. The final purification of each compound was achieved on a RP C<sub>18</sub> column (3 × 30 cm) equilibrated with CH<sub>3</sub>OH/H<sub>2</sub>O (1:1). The compounds were eluted in CH<sub>3</sub>OH/H<sub>2</sub>O (3:1) after the column had been washed with the equilibration solvent (250 mL).

The ratio of sample to RP packing material was approximately 1:100, whereas for silica gel it was 1:250. The pH of all the samples was adjusted to 3.5 prior to fractionation on Amberlite XAD-2 and RP C<sub>18</sub> columns. For samples of 50 mg or less final purification was accomplished using Sep-Pak C<sub>18</sub> cartridges.

## RESULTS AND DISCUSSION

In the present study the fundamental extraction and purification steps reported previously (Gelderblom et al., 1988a) were used, with minor modifications, for the isolation of FB<sub>1</sub>, FB<sub>2</sub>, FB<sub>3</sub>, and FB<sub>4</sub> and other structurally related compounds. Chemical structures of the newly discovered fumonisins B<sub>3</sub> and B<sub>4</sub> are given in Figure 1.

Extraction of the fumonisins was carried out using CH<sub>3</sub>OH/H<sub>2</sub>O (3:1) after the culture material had been extracted with ethyl acetate to remove lipid-soluble material. The latter step was very effective in that, although approximately 8% of material was recovered from the ethyl acetate extract, no fumonisin mycotoxins were extracted (Table I). Subsequent extraction with aqueous CH<sub>3</sub>OH resulted in an 81.3% recovery of FB<sub>1</sub> from the culture material (Table I). The recovery of the analytical technique for the quantitation of FB<sub>1</sub> in fungal cultures, using the same extraction solvent, is on the order of 85% (Alberts et al., 1990). In contrast, a far lower yield of FB<sub>2</sub> (59.8%) was obtained which could partly be ascribed to the extraction solvent used, as it was selected for the isolation and quantification of FB<sub>1</sub> (Alberts et al., 1990). From this study it would appear that aqueous methanol is not a good extraction solvent for FB<sub>2</sub>, and further investigation is necessary.

A high-percentage recovery of both FB<sub>1</sub> (98%) and FB<sub>2</sub> (97.5%) was obtained from the aqueous phase after the CHCl<sub>3</sub> partitioning step, while a low (1.1-fold) purification was achieved. The aim of the solvent-partitioning step is, as in the case with the ethyl acetate extraction, to remove lipid-soluble material that could interfere during the Amberlite XAD-2 column purification step.

Chromatographic separation on the Amberlite XAD-2 column was performed by a stepwise increase in the CH<sub>3</sub>OH content of the eluting solvent (Chart I). Fractionation on this column is based on group separation. The bulk (>94%) of the fumonisins was recovered in the final methanol eluate (Tables II and III), while a high degree of purification was achieved (12.1-fold for FB<sub>1</sub>, 8.7-fold

Table II. Quantification of FB<sub>1</sub> and FB<sub>2</sub> during Successive Column Purification Steps

purification step	sample wt, <sup>a</sup> g	fumonisin wt, <sup>a</sup> g	recovery, <sup>b</sup> %	fumonisin concn, %	purification <sup>c</sup>	
					degree	fold
solvent partitioning	125.0 ± 32.6	FB <sub>1</sub> 2.64 ± 0.28 FB <sub>2</sub> 1.23 ± 0.17	98 97.5	2.1 1.0		1.1
XAD (CH <sub>3</sub> OH)	(i) 10.0 ± 0.6 (ii) 13.3 ± 0.1	FB <sub>1</sub> 2.55 ± 0.18 FB <sub>2</sub> 1.16 ± 0.16	98 94	25.5 8.7	12.1 8.7	12.5 9.6
silica gel (6:3:1)	(i) 4.75 ± 0.22 (ii) 5.60 ± 0.22	FB <sub>1</sub> 1.90 ± 0.15 FB <sub>2</sub> 0.68 ± 0.03	74.5 58	40.0 12.1	1.6 1.4	26 22
silica gel (55:36:8:1)	(i) 2.14 ± 0.32 (ii) 1.30 ± 0.07	FB <sub>1</sub> 1.30 ± 0.35 FB <sub>2</sub> 0.49 ± 0.14	68 72	60.7 37.7	1.5 3.1	58 96
RP (C <sub>18</sub> )	(i) 1.33 ± 0.44 (ii) 0.50 ± 0.04	FB <sub>1</sub> 1.20 ± 0.29 FB <sub>2</sub> 0.46 ± 0.01	91 94	90.2 92.0	1.5 2.4	94 250

<sup>a</sup> All values represent means ± standard deviation calculated from performing each purification in triplicate. The quantitation of the recoveries of FB<sub>1</sub> and FB<sub>2</sub> was done in two separate triplicate series. <sup>b</sup> Percentage fumonisin recovered during each purification step. <sup>c</sup> Degree of purification represents the increase in fumonisin concentration achieved by each purification step, while the fold represents the extent of purification of each fumonisin after the solvent-partitioning step.

Table III. Quantification of FB<sub>2</sub> and FB<sub>3</sub> during Successive Column Purification Steps

purification step	sample wt, <sup>a</sup> g	fumonisin wt, <sup>a</sup> g	recovery, <sup>b</sup> %	fumonisin concn, %	purification <sup>c</sup>	
					degree	fold
solvent partitioning	125 ± 32.6	FB <sub>2</sub> 1.23 ± 0.17 FB <sub>3</sub> 0.62 ± 0.08		1.0 0.5		
XAD (CH <sub>3</sub> OH)	13.3 ± 0.1	FB <sub>2</sub> 1.16 ± 0.16 FB <sub>3</sub> 0.61 ± 0.08	(a) 94 (b) 98	8.7 4.6	8.7 9.2	9.6 9.6
silica gel (6:3:1)	3.24 ± 0.17	FB <sub>2</sub> 0.49 ± 0.06 FB <sub>3</sub> 0.52 ± 0.08	(a) 42 (b) 85	15.1 16	1.7 3.5	39 39
XAD (CH <sub>3</sub> OH/CH <sub>3</sub> CN)	2.16 ± 0.20	FB <sub>2</sub> 0.46 ± 0.09 FB <sub>3</sub> 0.51 ± 0.08	(a) 94 (b) 98	21.3 23.6	1.4 1.5	58 58
silica gel (55:39:5:1)	1.99 ± 0.10	FB <sub>2</sub> 0.35 ± 0.02 FB <sub>3</sub> 0.36 ± 0.03	(a) 76 (b) 71	17.6 18.1	0.8 0.8	63 63
RP (C <sub>18</sub> )	(a) 0.37 ± 0.02 (b) 0.38 ± 0.02	FB <sub>2</sub> 0.33 ± 0.03 FB <sub>3</sub> 0.36 ± 0.02	(a) 94 (b) 99	89.2 94.7	5.1 5.2	(a) 338 (b) 328

<sup>a</sup> All values represent means ± standard deviation calculated from performing each purification step in triplicate. <sup>b</sup> Percentage fumonisin recovered during each purification step. <sup>c</sup> Degree of purification represents the increase in fumonisin concentration achieved by each purification step, while the fold represents the extent of purification of each fumonisin after the solvent-partitioning step. Quantitation of FB<sub>2</sub> and FB<sub>3</sub> eluted from the XAD-2 column was performed on two different CH<sub>3</sub>OH eluates (a and b).

for FB<sub>2</sub>, Table II; 9.2-fold for FB<sub>3</sub>, Table III). This chromatographic procedure is therefore by far the most effective purification step for removing residual material other than the fumonisins, although no separation between the individual fumonisins is obtained. In addition, polar components that would interfere with the subsequent purifications on silica gel were effectively removed from the aqueous extract.

The most effective separation between the different fumonisins was obtained on silica gel by using two different mobile phases. Both column separations resulted in approximately 70% recovery of FB<sub>1</sub> (Table II). During the first silica gel column separation, using CHCl<sub>3</sub>/CH<sub>3</sub>OH/CH<sub>3</sub>OOH (6:3:1) as the eluant, most of the pigmented material (Steyn et al., 1979) present in culture extracts of *F. moniliforme* MRC 826 was separated from FB<sub>1</sub>, FB<sub>2</sub>, and FB<sub>3</sub>. When this purification step was omitted, some of the pigments were present in the final FB<sub>1</sub> preparation. This purification step did not completely separate FB<sub>1</sub> from FB<sub>2</sub> and FB<sub>3</sub>, but as the aim of this study was directed at the purification of FB<sub>1</sub>, the fractions were combined separately, thus yielding two main fractions. One fraction (fractions 51–90) contained the bulk of the FB<sub>1</sub> as well as about 50% of the FB<sub>2</sub> and far less FB<sub>3</sub>, while the other fraction (fractions 30–50) consisted mainly of FB<sub>3</sub> and some FB<sub>2</sub> (Chart I). The recovery of FB<sub>3</sub> from the first silica gel column was 85% (Table III). The FB<sub>2</sub> was recovered quantitatively from the first silica gel column with 58% being obtained from the fractions containing FB<sub>1</sub> (Table II) and 42% from those containing FB<sub>3</sub> (Table III). The lower recovery for FB<sub>1</sub> (74.5%, Table II) can be explained by the long elution time and its tendency to streak from silica gel.

The advantage of the second silica gel column purification step in the isolation of FB<sub>1</sub> was that FB<sub>1</sub> was completely separated from FB<sub>2</sub> as shown by TLC. However, the material recovered in the combined FB<sub>1</sub> fractions contained much methanol-insoluble material, presumably salts, which only dissolved in aqueous methanol at a pH below 3.5. The subsequent RP C<sub>18</sub> column step was therefore aimed mainly at removing the inorganic material while the gradient elution of FB<sub>1</sub> also removed some yellow pigments that coeluted with FB<sub>1</sub> from the second silica gel column. A high recovery (91%) and relatively low purification (1.5-fold) of FB<sub>1</sub> was achieved in this chromatographic purification step (Table II).

Although the present method was developed mainly to purify FB<sub>1</sub>, it became clear that, with minor modifications in the sequence of the columns and eluants used, FB<sub>2</sub> and FB<sub>3</sub> could also be purified without any major difficulties. Approximately 50% of the FB<sub>2</sub> was purified to over 90% purity by applying the same column purification procedures as for FB<sub>1</sub> (Chart I; Table II). The introduction of another Amberlite XAD-2 column in the purification of the main FB<sub>2</sub>/FB<sub>3</sub> sample (containing the other half of the FB<sub>2</sub>) was to effect the removal of CH<sub>3</sub>OH-insoluble material which coeluted with this fraction from the first silica gel column. As discussed earlier, this material elutes mainly with FB<sub>1</sub> from the second silica gel column. The use of CH<sub>3</sub>CN in place of CH<sub>3</sub>OH during the second Amberlite XAD-2 chromatographic step was necessary because of the lower polarity of FB<sub>2</sub> and FB<sub>3</sub> in comparison to that of FB<sub>1</sub> (Table IV). A very high recovery of FB<sub>2</sub> (94%) and FB<sub>3</sub> (98%) from the Amberlite XAD-2 column was achieved (Table III). FB<sub>2</sub> and FB<sub>3</sub> coeluted from both the second Amberlite XAD-2 column and the subsequent

**Table IV. Chemical and Chromatographic Characteristics of the Fumonisin and Structurally Related Derivatives**

compound	$R_f$ value <sup>a</sup>	FB <sub>1</sub>	FB <sub>2</sub>
FB <sub>1</sub>	0.23 (0.15) <sup>b</sup>		
FB <sub>2</sub>	0.30 (0.20)		
FB <sub>3</sub>	0.30 (0.26)		
FB <sub>4</sub>	0.37 (0.31)		
MMEs(1) <sup>c</sup>	0.35 (0.27)	monomethyl ester	
MMEs(2) <sup>c</sup>	0.42 (0.34)		monomethyl ester
DMEs(1) <sup>c</sup>	0.48 (0.42)	dimethyl ester	
DMEs(2) <sup>c</sup>	0.54 (0.46)		dimethyl ester
FA <sub>1</sub>	0.38 (0.71)	<i>N</i> -acetyl derivative	
FA <sub>2</sub>	0.46 (0.76)		<i>N</i> -acetyl derivative

<sup>a</sup> Silica gel TLC using CHCl<sub>3</sub>:CH<sub>3</sub>OH:H<sub>2</sub>O:CH<sub>3</sub>COOH (55:36:8:1) as developing solvent. <sup>b</sup> Values in parentheses represent  $R_f$  values with CHCl<sub>3</sub>:CH<sub>3</sub>OH:CH<sub>3</sub>COOH (6:3:1) as developing solvent. <sup>c</sup> MMEs, monomethyl esters; DMEs, dimethyl esters.

silica gel column but were finally separated on RP C<sub>18</sub>. The latter step is the only chromatographic procedure which effectively separates FB<sub>2</sub> from FB<sub>3</sub>. As indicated in Chart I, FB<sub>4</sub> is also purified during these chromatographic procedures.

Fumonisin B<sub>1</sub>, B<sub>2</sub>, and B<sub>3</sub> are the major naturally produced fumonisins in corn cultures of *F. moniliforme* strain MRC 826. FB<sub>4</sub>, FA<sub>1</sub>, and FA<sub>2</sub> are produced only in minor quantities. A major disadvantage of the main separation between the different fumonisins on silica gel is the fact that several structurally related compounds (MMEs and DMEs) are formed from FB<sub>1</sub>, FB<sub>2</sub>, and probably FB<sub>3</sub> as a result of the presence of CH<sub>3</sub>OH and CH<sub>3</sub>COOH in the mobile phase (Table IV; Chart I). The carboxylic acid group involved in the formation of each ester is not known. The formation of the MMEs complicates the purification of FB<sub>2</sub> and FB<sub>3</sub> especially during their fractionation on silica gel as they eluted in the same region, as indicated by the  $R_f$  values obtained by TLC (Table IV). The same problem is encountered on the C<sub>18</sub> columns where good separation was obtained between FB<sub>1</sub>, FB<sub>2</sub>, and FB<sub>3</sub> but not between FB<sub>3</sub> and the MMEs of FB<sub>1</sub> and FB<sub>2</sub>, respectively. The purification of FB<sub>4</sub>, which appears to be produced only in minor quantities by the fungus, is hampered by the same difficulties. No problems were encountered in the purification of the dimethyl ester derivatives of both FB<sub>1</sub> and FB<sub>2</sub> as they are eluted much earlier from silica gel (higher  $R_f$  values).

FA<sub>1</sub> and FA<sub>2</sub> were obtained after the first silica gel column separation, and NMR spectral analyses indicated that these compounds are the *N*-acetyl derivatives of FB<sub>1</sub> and FB<sub>2</sub>, respectively. These compounds have previously been reported as metabolites produced by *F. moniliforme* in culture (Bezuidenhout et al., 1988). This finding is confirmed in the present study as there is no indication that either FB<sub>1</sub> or FB<sub>2</sub> is converted into these compounds in the presence of CH<sub>3</sub>COOH even when incubated at fairly high temperatures (60 °C).

Although the method described in this paper is not optimal with respect to the quantities of FB<sub>1</sub>, FB<sub>2</sub>, and FB<sub>3</sub> that are purified, it does provide ample amounts of these mycotoxins for biological evaluation in experimental animals. The high concentration of FB<sub>1</sub> present in culture material of this specific fungal isolate facilitates the purification of FB<sub>1</sub> over that of FB<sub>2</sub> and FB<sub>3</sub> as indicated by the much greater ease (lower purification fold; Table II) with which pure FB<sub>1</sub> can be obtained. At present the fumonisin production profiles of other isolates of *F. mo-*

*niliforme* are under investigation in an attempt to select those isolates which will allow optimal purification of each individual fumonisin. Other solvent systems for the purification of the fumonisins on silica gel are also under investigation to prevent the formation of the structurally related compounds, thereby increasing the final yield of the pure fumonisins.

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**Registry No.** FA<sub>1</sub>, 117415-48-2; FA<sub>2</sub>, 117415-47-1; FB<sub>1</sub>, 116355-83-0; FB<sub>1</sub> monomethyl ester, 136458-38-3; FB<sub>1</sub> dimethyl ester, 136458-39-4; FB<sub>2</sub>, 116355-84-1; FB<sub>2</sub> monomethyl ester, 136520-22-4; FB<sub>2</sub> dimethyl ester, 136458-40-7; FB<sub>3</sub>, 136379-59-4; FB<sub>4</sub>, 136379-60-7.